

# Supplemental Material to “Counterpropagating dual-trap optical tweezers based on linear momentum conservation”.

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obtained by digestion and PCR amplification of the phage  $\lambda$  genome. All experiments were performed in a microfluidics chamber formed by two coverslips interspaced with parafilm. The chamber has three channels: a central one where experiments are carried out and two (upper and lower) channels that are connected to the central one by two dispenser tubes. Anti-dig coated beads were first incubated with the molecule of interest and then introduced in the microfluidics chamber through one of the dispenser tubes. Once the anti-dig coated bead was trapped a streptavidin coated bead was introduced through a second dispenser tube and trapped in the second trap. The connection was then formed directly inside the microfluidics chamber. All experiments on DNA tethers were performed in PBS buffer solution at pH 7.4, 1M NaCl, at 25° C. This buffer solution was found to greatly reduce the nonspecific adsorption of DNA on silica, allowing the use of commercial beads from Kisker Biotech without any specific preparation or coating. We dissolved 1mg/ $\mu$ l Bovine Serum Albumine in the buffer in order to reduce silica-silica interactions.

## II. RECONSTRUCTION OF A COARSE FREE ENERGY LANDSCAPE

The mechanical folding and unfolding of nucleic acid hairpins is commonly described in terms of a reaction coordinate and of the corresponding free energy landscape. When subject to force, the end-to-end distance of the molecule along the force axis is an adequate reaction coordinate for the folding-unfolding reaction pathway. For a given applied force  $f$ , a single kinetic pathway for the unfolding and folding reactions is usually considered, characterized by a single transition state (TS). The TS is the highest free-energy state encountered along the reaction coordinate and determines the kinetics of the folding-unfolding reaction. The model involves four parameters: the free energy of folding at zero force,  $\Delta G = G_F - G_U$ , the height of the kinetic barrier  $B_0$ , defined as the free energy difference between the transition state and the folded (F) state extrapolated to zero force, and the distances  $x_F$  and  $x_U$  along the reaction coordinate that separates the transition state from states  $F$  and  $U$  respectively; the total distance along the reaction coordinate being  $x_m = x_F + x_U$ . Under an applied force the free energy landscape is tilted along the reaction coordinate and the free-energy difference  $\Delta G$  and the barrier  $B_0$  change accordingly. To a first approximation,  $\Delta G$  and  $B_0$  depend linearly on the force whereas  $x_F$  and  $x_U$  are taken force-independent. Hence the

reaction rates are given by

$$k_{F \rightarrow U} = k_0 e^{-\beta(B - G_F - x_{FU}f)} = k_m e^{\beta x_{FU}f} \quad (1)$$

$$k_{U \rightarrow F} = k_0 e^{-\beta(B - G_U + x_{UF}f)} = k_m e^{\beta(\Delta G_0 - x_{UF}f)}, \quad (2)$$

and  $k_m = k_0 e^{-\beta B_0}$  is an effective attempt rate. The free-energy difference between state U and F under the given force  $f$  is given by,

$$\Delta G(f) = -k_B T \log \left( \frac{k_{F \rightarrow U}}{k_{U \rightarrow F}} \right) = \Delta G_0 - x_m f. \quad (3)$$

with  $\Delta G_{FU}(f) = G_F(f) - G_U(f)$ . The four parameters describing the free-energy landscape can therefore be reconstructed from the kinetic rates measured at different forces.

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- [1] N. Forns, S. de Lorenzo, M. Manosas, K. Hayashi, J.M. Huguet and F. Ritort, Biophysical Journal **100**, 1765 (2011)